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10 Marking scheme: End-of-chapter test

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1 $Q = CV = 470 \times 10^{-6} \times 9.0 = 4.2 \times 10^{-3} \text{ C}$ [1] 2 $V = Q/C = 0.033/2200 \times 10^{-6} = 15 \text{ V}$ [1] 3 $C = Q/V = 2.0/5000 = 4.0 \times 10^{-4} = 400 \mu\text{F}$ [1] 4 $W = 1/2 CV^2 = 0.5 \times 470 \times 10^{-6} \times 12^2 = 0.034 \text{ J}$ [1] 5 $W = 1/2 QV = 0.5 \times 1.5 \times 10^{-3} \times 50 = 0.0375 \text{ J}$ [1] 6

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the product collected at the end of the column. Figure 10.7 a State one way that amylase could have been immobilised. [1] b One advantage of using immobilised enzymes in an industrial process is that the product should not be contaminated with the enzyme, meaning that there is no cost involved in purifying the product.

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4 End-of-chapter test

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Cambridge End Of Chapter Test 562.1 g 10.3 g 27.6 g. Number of moles. $12.0 \text{ g} / 62.1 \text{ g mol}^{-1} = 5.18 \times 10^{-2} \text{ mol}$ $10.3 \text{ g} / 16.0 \text{ g mol}^{-1} = 1.73 \times 10^{-2} \text{ mol}$ [1] Relative number of atoms $5.18 \times 10^{-2} \text{ mol} / 1.73 \times 10^{-2} \text{ mol} = 3$ $10.3 \text{ g} / 1.73 \text{ g mol}^{-1} = 6$ $1.73 \text{ g} / 1.73 \text{ g mol}^{-1} = 1$. 10 Marking scheme: End-of-chapter test - Cambridge Essentials 13 Marking scheme: End-of-chapter test. 1 a. $\text{C}_n\text{H}_{2n+1}\text{OH}$ [1] cButan-1-ol and Page 5/22

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