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1 self-ionization of water

occurs when two water

molecules produce a

hydronium (H_3O^+) and a

hydroxide (OH^-) ion 2 H_2O

$\text{H}_3\text{O}^+ + \text{OH}^-$ The Page 11/18

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occurs when two water
molecules produce a
hydronium (H_3O^+) and a
hydroxide (OH^-) ion $2 \text{H}_2\text{O}$
 $\text{H}_3\text{O}^+ + \text{OH}^-$ The ionization
constant (K_w) of water is:
 $K_w = [\text{H}_3\text{O}^+] [\text{OH}^-] = 1.0 \times$
 10^{-14} M Acidic, Basic, &
Neutral IF $[\text{H}_3\text{O}^+] >$

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CHAPTER 15 REVIEW Acid-Base
Titration and pH SECTION 2
SHORT ANSWER Answer the
following questions in the

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~~Review Answers~~ 1. Below is a pH curve from an acid-base titration. On it are labeled three points: X, Y, and Z.
a. Which point represents the equivalence point? b. At which point is there excess acid in the system? c.

~~15 Acid Base Titration and pH~~

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mhardiell. Terms in this set
(10) Self-ionization of
water. two water molecules
produce a hydroniumion and a
hydroxide ion by transfer of
a proton. pH. the negative
of the common logarithm of
the hydronium ion
concentration. pOH.

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SHORT ANSWER Answer the

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Following questions in the space provided. 1. Below is a pH curve from an acid-base titration. On it are labeled three points: X, Y, and Z.

_____ a.

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two or more substances in a
single phase. Soluble
describes a substance as

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capable of being dissolved.
Solvent is the dissolving
medium in a solution. Solute
is the substance that is ...

The ocean has absorbed a significant portion of all human-made carbon dioxide emissions. This benefits human society by moderating the rate of climate change, but also causes unprecedented changes to ocean chemistry. Carbon dioxide taken up by the ocean decreases the pH of the water and leads to a suite of chemical changes collectively known as ocean acidification. The long term

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consequences of ocean acidification are not known, but are expected to result in changes to many ecosystems and the services they provide to society.

Ocean Acidification: A National Strategy to Meet the Challenges of a Changing Ocean reviews the current state of knowledge, explores gaps in understanding, and identifies several key findings. Like climate change, ocean acidification is a growing global problem that will intensify with continued CO₂ emissions and has the potential to change marine ecosystems and affect benefits to society. The federal government has taken

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positive initial steps by developing a national ocean acidification program, but more information is needed to fully understand and address the threat that ocean acidification may pose to marine ecosystems and the services they provide. In addition, a global observation network of chemical and biological sensors is needed to monitor changes in ocean conditions attributable to acidification.

Long considered the standard for honors and high-level mainstream general chemistry courses, PRINCIPLES OF MODERN CHEMISTRY continues

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Coordination compounds and coordination polymers, cluster compounds, organometallic compounds, inorganic polymers, and nonstoichiometric compounds. Section three elaborates on the synthetic chemistry of five important classes of inorganic functional materials, namely, ordered porous materials, carbon materials, advanced ceramic materials, host-guest materials, and hierarchically structured materials. Section four consists of four chapters where the synthesis of functional inorganic aggregates is discussed, giving special attention to

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the growth of single crystals, assembly of nanomaterials, and preparation of amorphous materials and membranes. The new edition's biggest highlight is Section five where the frontier in inorganic synthetic chemistry is reviewed by focusing on biomimetic synthesis and rationally designed synthesis. Focuses on the chemistry of inorganic synthesis, assembly, and organization of wide-ranging inorganic systems Covers all major methodologies of inorganic synthesis Provides state-of-the-art synthetic methods Includes real examples in

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chemistry students per year and creates content to offer more in-depth explanation in areas where she knows they have problems. Continuing in the Burdge tradition, the fourth edition maintains an outstanding art program, a consistent problem-solving approach, interesting applications woven throughout the chapters, and a wide range of end-of-chapter problems.

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representative inorganic materials are selected for discussion of their preparation and assembly, including porous, advanced ceramic, amorphous- and nano-materials, inorganic membranes, synthetic crystals and advanced functional materials. The last part of the book, which is also its distinct feature, addresses the frontiers of inorganic synthesis and preparative chemistry. These final two chapters introduce the two emerging synthetic areas. Included are approximately 3000 references, a large proportion of which are from the recent decade. Focuses

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various compounds and
describes all inorganic
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chemical synthesis have led

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chemical engineering and current world-wide activities. Requirements with respect to needs of the chemical industry have been included. Chemists, chemical engineers, biotechnologists, process engineers, microsystem technologists in the chemical and pharmaceutical industry and academia, as well as manufacturers of analytical instruments, will find this book a state-of-the-art review of this extremely interesting and rapidly developing field.

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with an emphasis on 20th century developments and more recent findings.

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